

Describe each of the following compounds as soluble, or of low solubility

a.	CuBr	low
b.	CuCl <sub>2</sub>	
c.	NaOH	
d.	Na <sub>3</sub> PO <sub>4</sub>	
e.	AgOH	low
f.	MgS	
g.	Sr(OH) <sub>2</sub>	
h.	PbSO <sub>4</sub>	low
i.	CdSO <sub>4</sub>	
j.	Rb <sub>2</sub> SO <sub>4</sub>	
k.	CuOH	low
l.	BaSO <sub>4</sub>	low
m.	NH <sub>4</sub> OH	
n.	Fr <sub>2</sub> S	
o.	CuSO <sub>4</sub>	
p.	CrS	low
q.	(NH <sub>4</sub> ) <sub>2</sub> Cr <sub>2</sub> O <sub>7</sub>	
r.	K <sub>2</sub> CrO <sub>4</sub>	
s.	NaCl	
t.	KNO <sub>3</sub>	
u.	NH <sub>4</sub> NO <sub>3</sub>	
v.	HCOOH	
w.	PbCO <sub>3</sub>	low
x.	PbNO <sub>3</sub>	
y.	Na <sub>2</sub> SO <sub>3</sub>	